STEREOSPECIFIC CYCLOPROPENIC SYNTHESIS OF <u>CIS</u>-CHRYSANTHEMIC METHYL ESTER.2. THE BY-PASSING DIAZOALKANE WAY (1).

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A stereospecific synthesis of cis-chrysanthemate by catalytic hydrogenation of a cyclopropene is described. The required precursor, a 3H-pyrazole, is obtained here via a novel dianion reaction and MnO_2 dehydrogenation of a Δ^2 -pyrazoline, instead of the more cumbersome diazoaddition method.

The powerful natural insecticides from pyrethrum are all esters of \underline{trans} -chrysanthemic or pyrethric acids. It was however seen that unnatural \underline{cls} -chrysanthemic esters can show higher activities (2). We reported recently on one of the very few total stereospecific syntheses of \underline{cls} -chrysanthemic methyl ester in which the key step involved the selective hydrogenation of the corresponding cyclopropenic ester (3) :



This synthesis, although very simple and fortunately devoid of complications, suffers nevertheless from some drawbacks, the main of which being the use of the highly unstable 2-diazopropane (DAP). This paper describes now a new cyclopropenic synthesis of <u>cis</u>-chrysanthemic ester which does not necessitate an acetylenic starting ester nor an unstable diazoalkane. Furthermore, no sophisticated selective hydrogenation step is required.

The easiest way to reach to a cyclopropenic ester being still the photolysis of a 3H-pyrazole, the problem was to obtain the latter otherwise than by a 1,3-dipolar cycloaddition reaction. With this goal in mind, we found that a number of 5,5-dimethyl Δ^2 -pyrazolines can rapidly be dehydrogenated in high yields to 3H-pyrazoles with manganese dioxide (4,5). The ease of this conversion is however strongly dependent from the substitution pattern and we had, at first, to check the feasibility of the reaction in this particular case. The cycloaddition of 2-diazopropane to methyl δ -methylsorbate leads, after acidic treatment, to a mixture of the Δ^2 -pyrazolines <u>1</u> and <u>2</u> (4,6). The reversed regiolsomer <u>2</u>, is cleanly oxidized by MnO₂ into the known 3H-pyrazole <u>3</u> (3),

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whereas the isomeric structure <u>1</u> is transformed into a mixture of several unidentified compounds.



The required Δ^2 -pyrazoline, bearing the ester group in position 4, has now to be obtained without a diazoalkane. This is generally achieved most simply by condensation of hydrazine and an α,β -unsaturated ketone (7). As the isobutenylic double bond can be formed in the last step, and in order to escape side reactions during the condensation step with hydrazine, we started from the tertiary alcoholic enone ester <u>6</u> conveniently prepared from the known compound <u>4</u> (8) by a novel diamion reaction (4,9) :



a: 1,1 equiv. NaH , THF , 65°C b 1,1 equiv Buli , 0°C , 15 min

c 3 equiv. CH_3COCH_3 , r.t., 15 min d $10g A1_2O_3$ basic form for 1g crude 5, MeOH; 12 h 6: pale yellow liquid, isolated in 72 % yield from 4, by SiO₂ column chromatography (Hexane-Ether 1 1), IR v(OH) 3500 cm⁻¹ (broad), v(C=O) 1720 cm⁺² and 1685 cm⁻¹, v(C=C) 1630 cm⁻¹, NMR δ = 1,30 (6H,s) 1,96 (3H,s) 2,12 (3H,s) 2,74 (2H,s) 3,74 (3H,s) and 3,90 (1H, broad s)

The condensation with hydrazine hydrate has to be performed rather drastically, by heating in acetic acid as solvent, distilling finaly the acid under normal pressure, until the temperature of the mixture reaches 180°C. In these conditions the almost pure Δ^2 -pyrazoline 7 is obtained in nearly quantitative yield after simple dilution with water and extraction with chloroform. The following oxidation with MnO₂ (10) in methylene chloride is a rapid reaction, giving the sole 3H-pyrazole 8 (11) :



a. 1,1 equiv. N_2H_4 , H_2O , AcOH, 180°C b. 8g MnO₂ for 1g $\underline{7}$, CH_2C1_2 under Argon , r t. , 1 h 7 pale yellow liquid , IR broad absorption between 3300 and 3500 cm-

8. pale yellow liquid, isolated pure in 86 % yield from 6, by filtration on SiO₂, IR ν (OH) 3500 cm⁻¹, ν (C=O) 1710 cm⁻¹, C=C-C=N 1635 cm⁻¹, UV λ max = 262 nm (7600) and 368 nm (280), NMR δ = 1,30 (6H,s) 1,53 (6H,s) 3,32 (2H,s) 3,48 (1H,s) and 3,85 (3H,s).

The photolysis of the 3H-pyrazole $\underline{8}$ (12) leads then to the unique cyclopropene $\underline{9}$, which has not to be isolated and is directly hydrogenated catalytically over 5% Palladium on chaircoal. Rapid adsorption of the stoechiometric amount of hydrogen occurs giving the <u>cis</u>-cyclopropanic derivative <u>10</u>.



- 9 can be obtained in 95% yield by evaporation of the irradiation solvent, but not further purified; liquid, IR ν (OH) 3450 cm⁻¹ (broad), caracteristic cyclopropenic absorption at 1825 cm⁻¹, ν (C=O) 1680 cm⁻¹, NMR δ = 1,25 (6H,s) 1,30 (6H,s) 2,76 (2H,s) 3,82 (3H,s) and 4,02 (1H,s)
- 10 colourless liquid, isolated pure in 94% yield by filtration on SiO₂ (Hexane-Ether 2 1), IR $\nu(0H)$ 3400 cm^{-(broad)}, $\nu(C=0)$ 1705 cm⁻⁽⁾, NMR $\delta = 1,20$ (6H,s) 1,24 (6H,s) 3,62 (3H,s) other H as multiplets between 1,30 and 1,92 (5H).

The clean conversion of the pyrazole $\underline{8}$ into a cyclopropene is remarkable here in view of the fact that the reaction proceeds in this case through an alkylvinylcarbene with an alcoholic function in the same molecule (13).

The last step, i.e. the introduction of the double bond was best achieved with mesyle chloride in presence of an excess of triethylamine. In these conditions no lactonization is observed, in spite of the favourable <u>cis</u> configuration (14), and a 1:2 mixture of <u>cis</u>-chrysanthemic and <u>cis</u>-isochrysanthemic methyl esters is obtained (85%). Isomerization with sulfuric acid, a known procedure (15), gives then the sole <u>cis</u>-chrysanthemate <u>12</u> isolated after SiO₂ filtration (2% Ether in hexane) in 76% yield based on the 3H-pyrazole <u>8</u> (Overall yield from compound <u>4</u> : 47%). A second possibility, in fact the first we tried, is to transform, in contrary, the alcohol <u>10</u> into the lactone <u>11</u> (t. BuOK , bz, 80°C ; 1 h) which can in turn lead to <u>cis</u>-chrysanthemic acid (60% from the pyrazole) by the known procedure of J. Martel et al. (MgBr₂, pyridine ; 125°C ; 14 h) (16).



a. 3,5 equiv. Et $_3$ N, CH $_2$ Cl b 1,5 equiv MsCl; CH $_2$ Cl $_2$ -5°C, 20 min, rt, 12 h c. H $_2$ SO $_4$ cat , diox. 80°C, 4 h

This work was supported by the CNRS (ATP "Agrochimie", Décision n° 3513). We are

also grateful for financial support from Roussel-Uclaf.

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The indicated yields are corresponding to products which gave satisfactory microanalyses, IR (CHCl₃), UV (CH₃CN) and H^1 -NMR (CDCl₃/TMS) spectra

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(Received in UK 25 January 1982)